

Pogil Answer Key Oxidation And Reduction

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from Model 1 to support your answer. /Vâf Oxidation and Reduction . 3. In the space under each reaction in Model I, write the oxidation number for every atom. Divide the work among your group members. An example is shown here: 4Fe(s) + 302(g) 2Fe20 (s) 4. Identify any elements that changed oxidation number in the reactions in Model I. Connect ...

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beautiful green patina as it ages. Metals rust or corrode in the presence of air and water. Minerals (ionic. compounds) found in ore can be decomposed with the use of electricity to produce pure metals and. nonmetals. All of these reactions are examples of oxidation and reduction, otherwise known as redox. reactions.

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Oxidation Reduction Reactions- Answer Key. 4.51. If nitric acid is a strong oxidizing agent and zinc is a strong reducing agent, then zinc metal will probably reduce nitric acid when the two react; that is, N will gain electrons and the oxidation number of N must decrease.

Oxidation Reduction Reactions- Answer Key
Pogil Answer Key Oxidation And Reduction Pdf - â€¦ Oxidation and Reduction 3 Read This! The process of oxidation and reduction can be thought of as a transfer of electrons from one atom to another. Thus, one atom gives up electrons and the other atom gains them. As a result of this process, the oxidation numbers of both atoms change.

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